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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/542,618	07/18/2005	Megumu Nagasawa	2005_1140A	7292
513 7590 09/21/2009 WENDEROTH, LIND & PONACK, L.L.P. 1030 15th Street, N.W.,			EXAMINER	
			CREPEAU, JONATHAN	
Suite 400 East Washington, DC 20005-1503			ART UNIT	PAPER NUMBER
			1795	
			MAIL DATE	DELIVERY MODE
			09/21/2009	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

		Application No.	Applicant(s)			
Office Asticus Occurrence		10/542,618	NAGASAWA ET AL.			
	Office Action Summary	Examiner	Art Unit			
		Jonathan Crepeau	1795			
Period fo	The MAILING DATE of this communication app or Reply	pears on the cover sheet with the c	orrespondence address			
WHIC - Exter after - If NC - Failu Any	ORTENED STATUTORY PERIOD FOR REPLY CHEVER IS LONGER, FROM THE MAILING DATE in an analysis of 37 CFR 1.1. SIX (6) MONTHS from the mailing date of this communication. It is period for reply is specified above, the maximum statutory period or reto reply within the set or extended period for reply will, by statute reply received by the Office later than three months after the mailing and patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be tin will apply and will expire SIX (6) MONTHS from the cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).			
Status						
1)⊠	Responsive to communication(s) filed on <u>26 M</u>	lav 2009				
-		action is non-final.				
3)	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is					
- , 	closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.					
Dispositi	on of Claims					
4)🛛	4) Claim(s) 1 and 10-13 is/are pending in the application.					
,	4a) Of the above claim(s) is/are withdrawn from consideration.					
	5) Claim(s) is/are allowed.					
	6)⊠ Claim(s) <u>1 and 10-13</u> is/are rejected.					
· ·	Claim(s) is/are objected to.					
	Claim(s) are subject to restriction and/o	r election requirement.				
Applicati	on Papers					
9) The specification is objected to by the Examiner.						
•	10) ☐ The drawing(s) filed on is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.					
,	Applicant may not request that any objection to the					
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).						
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.						
Priority ι	ınder 35 U.S.C. § 119					
a)	Acknowledgment is made of a claim for foreign All b) Some * c) None of: 1. Certified copies of the priority document 2. Certified copies of the priority document 3. Copies of the certified copies of the priority document application from the International Bureausee the attached detailed Office action for a list	s have been received. s have been received in Applicati rity documents have been receive u (PCT Rule 17.2(a)).	on No ed in this National Stage			
2) Notice (3) Inform	t(s) e of References Cited (PTO-892) e of Draftsperson's Patent Drawing Review (PTO-948) mation Disclosure Statement(s) (PTO/SB/08) r No(s)/Mail Date	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal F 6) Other:	ate			

DETAILED ACTION

Response to Amendment

1. This Office action addresses claims 1 and newly added claims 10-13. Claims 1, 10, and 11 are rejected for substantially the reasons of record over WO '215, and claims 12 and 13 are newly rejected under 35 USC 103 as necessitated by amendment. Accordingly, this action is made final.

Claim Rejections - 35 USC § 103

2. Claims 1 and 10 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO 01/43215. Abe et al (U.S. Pre-Grant Publication No. 2003/0113611) is taken as an English equivalent of WO '215 herein.

Abe et al. '611 teach a fuel cell comprising electrodes and a proton exchange electrolyte membrane. In [0031] and [0032], the reference discloses that the electrodes each comprise an electroconductive porous substrate, an inorganic catalyst, polyaniline doped with a polymer sulfonic acid (proton acid), and a proton exchangeable resin. The inorganic catalyst can be platinum (see [0065]). Regarding claim 10, the polymer sulfonic acid can be phenolsulfonic acid novolac resin (see [0013]). Regarding claim 1, CO may be supplied to the anode with hydrogen (see [0007]).

Abe et al. do not teach that the polymeric acid (dopant) has an ion exchange capacity of 1.6 mg/eq or more, as recited in claim 1.

However, the invention as a whole would have been obvious to one of ordinary skill in the art at the time the invention was made because the artisan would be motivated to use a sulfonic acid polymer with a high proton conductivity in the electrodes of Abe et al. It has been held that the discovery of an optimum value of a result effective variable in a known process is ordinarily within the skill of the art. *In re Boesch*, 205 USPQ 215 (CCPA 1980). In this case, the use of a polymer having a high ion-exchange capacity would have more active proton sites available for the electrode oxidation-reduction reactions, as discussed in [0014] of Abe et al.

3. Claims 11 and 12 are rejected under 35 U.S.C. 103(a) as being unpatentable over WO '215 as applied to claims 1 and 10 above, and further in view of Fuglevand et al (U.S. Patent 6,218,035).

Abe et al. does not expressly teach that the proton acid polymer is crosslinked, as recited in claims 11 and 12.

In column 10, line 11, Fuglevand et al. teach a fuel cell comprising a crosslinked copolymer comprising sulfonic acid groups in the anode and cathode electrodes.

Therefore, the invention as a whole would have been obvious to one of ordinary skill in the art at the time the invention was made because a particular known technique (crosslinking ionically conductive polymers in electrodes) was recognized as part of the ordinary capabilities of one skilled in the art. Furthermore, crosslinking provides predictable results including

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increased strength. As such, it would have been obvious to crosslink the proton acid polymer of

Abe et al.

4. Claim 13 is rejected under 35 U.S.C. 103(a) as being unpatentable over WO '215 in view

of Fuglevand et al. as applied to claims 11 and 12 above, and further in view of Motogami et al

(U.S. Patent 5,116,541).

WO '215 and Fuglevand et al. do not expressly teach the novolac resin is crosslinked

with polyisocyanate, as recited in claim 13.

Motogami et al. is directed to an ion-conductive polymer electrolyte which may contain a

crosslinking agent such as an isocyanate compound (see abstract). In column 3, line 61 et seq.,

the reference teaches that the compound may be a polyisocyanate.

Therefore, the invention as a whole would have been obvious to one of ordinary skill in

the art at the time the invention was made because the artisan would be motivated to use a

polyisocyanate as the crosslinking agent of WO '215 as modified by Fuglevand et al. In column

4, line 67, Motogami et al. teach that "[w]hen an active hydrogen compound is crosslinked with

a crosslinking agent such as polyisocyanate and the like, a film superior in both of curability and

adhesion to the electrode can be obtained." Accordingly, the artisan would be motivated to use a

polyisocyanate as the crosslinking agent of WO '215 as modified by Fuglevand et al.

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Response to Arguments

5. Applicant's arguments filed May 26, 2009 have been fully considered but they are not persuasive. Applicants state that "Abe '611 does not disclose or suggest that the anode carries an ionic polymer sulfonic acid or a polymeric acid alone without the organic polymer." However, this argument is not persuasive because the present claims do not exclude such an organic polymer (i.e., polyaniline) from being present. In fact, claim 1 recites an additional "protonconductive ion exchange electrolytic polymer" present in the anode, although this is deemed to correspond to a different component in the Abe '611 anode. Applicants further state that "when the ionic polymer sulfonic acid of Abe '611 is used as a dopant for an organic polymer such as polyaniline, and forms a salt together with the polyaniline, the ionic polymer sulfonic acid can not provide any protons with a catalyst layer, in particular, in the neighborhood of noble metal particulate electrode catalyst," and accordingly, the polymer sulfonic acid acting as a dopant "has no function of a protonic acid." This argument is also not persuasive because the Applicant's assertion that the acid essentially loses its identity and forms a salt is not supported by the reference. The polymer sulfonic acid, although it is a "dopant," appears to retain its acidic character and is capable of supplying protons. See [0026], which expressly discloses a "powder of electroconductive polyaniline containing the polymer sulfonic acid as a dopant." Accordingly, the rejections as stated above are believed to be proper.

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Conclusion

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6. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, THIS ACTION IS MADE FINAL. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan Crepeau whose telephone number is (571) 272-1299. The examiner can normally be reached Monday-Friday from 9:30 AM - 6:00 PM EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan, can be reached at (571) 272-1292. The phone number for the organization where this application or proceeding is assigned is (571) 272-1700. Documents may be faxed to the central fax server at (571) 273-8300.

Information regarding the status of an application may be obtained from the Patent

Application Information Retrieval (PAIR) system. Status information for published applications

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/Jonathan Crepeau/ Primary Examiner, Art Unit 1795 September 19, 2009